



Metropolitan Water Reclamation District of Greater Chicago

**WELCOME
TO THE MAY EDITION
OF THE 2018
M&R SEMINAR SERIES**

BEFORE WE BEGIN

- **SAFETY PRECAUTIONS**
 - PLEASE FOLLOW EXIT SIGNS IN CASE OF EMERGENCY
 - AUTOMATED EXTERNAL DEFIBRILLATOR (AED) LOCATED OUTSIDE
- **PLEASE SILENCE CELL PHONES OR SMART PHONES**
- **A QUESTION AND ANSWER SESSION WILL FOLLOW PRESENTATION**
- **PLEASE FILL OUT THE EVALUATION FORM**
- **SEMINAR SLIDES WILL BE POSTED ON THE MWRD WEBSITE**
([www. MWRD.org](http://www.MWRD.org): Home Page ⇒ Reports ⇒ M&R Data and Reports ⇒ M&R Seminar Series ⇒ 2018 Seminar Series)
- **VIDEO STREAM OF THE PRESENTATION WILL BE AVAILABLE ON MWRD WEBSITE** (www.MWRD.org: Home Page ⇒ MWRDGC RSS Feeds)

MATTHEW J. HIGGINS, Ph.D.

- Matt Higgins is a currently Professor and Claire W. Carlson Chair in Environmental Engineering at Bucknell University. For the last 20 years, Matt has focused much of his research on biosolids issues including digestion, co-digestion, advanced digestion, conditioning and dewatering, mechanisms for production and control of odors in biosolids, and the reactivation and regrowth of indicators and pathogens in biosolids. He has collaborated significantly with both industry and municipalities, and his work focuses on understanding fundamental issues to solve real-world problems and support the industry's move toward more sustainable practices and a circular economy. His collaborative work with DC Water, AECOM, Brown and Caldwell and ARA Consult was recently awarded the Environmental Engineering Excellence Award from the American Academy of Environmental Engineers and the Excellence in Innovation Award from the Water Environment Research Foundation.
- Ph.D., Civil and Environmental Engineering, Virginia Tech
M.S. and B.S., Civil and Environmental Engineering, University of Maine

Understanding the Mechanisms of Dewatering to Explain the Negative Impacts of Biological Phosphorus Removal on Dewatering after Anaerobic Digestion

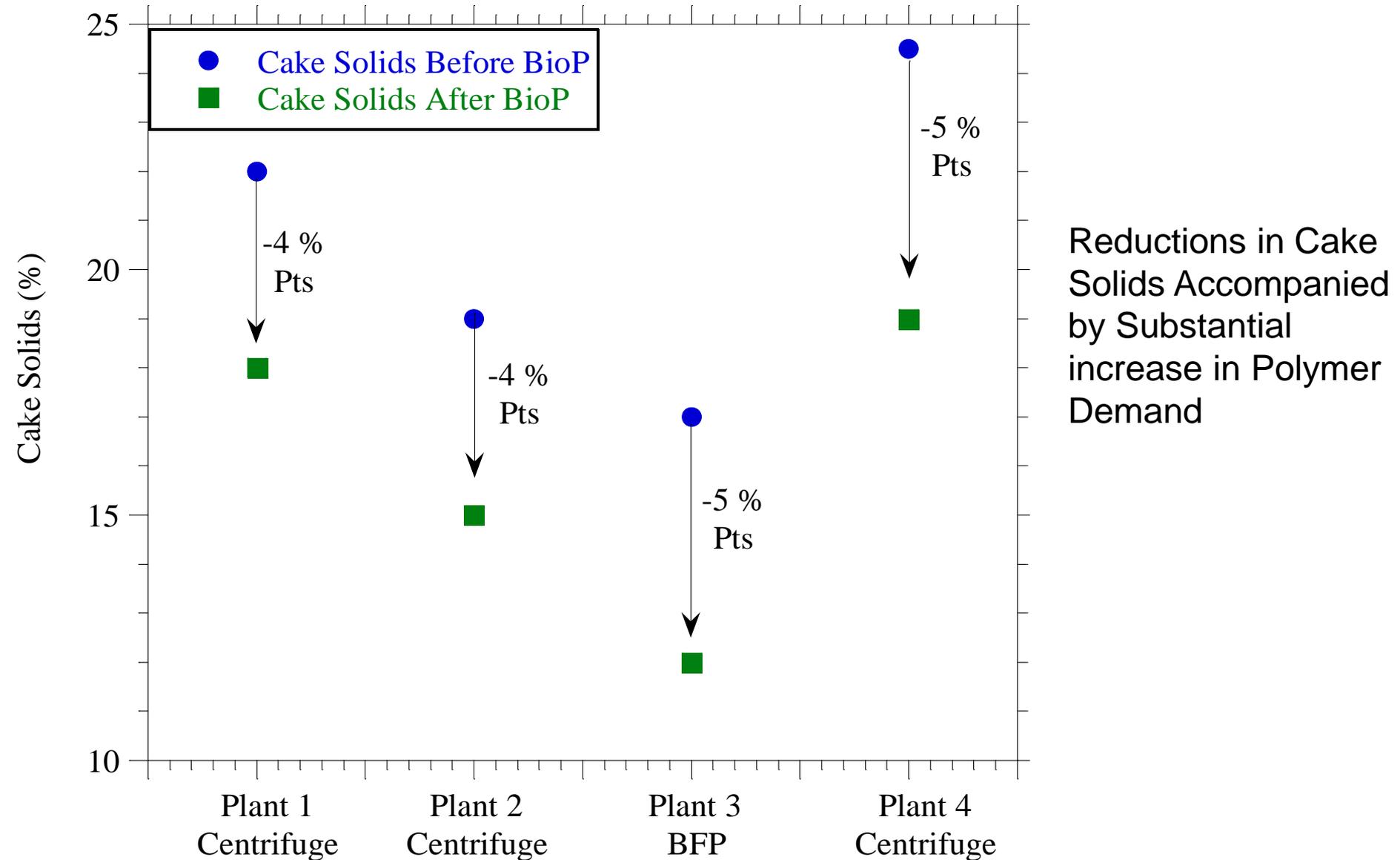
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Key Observation of Full-Scale BioP Plants

A number of plants with anaerobic digestion have observed a decrease in cake solids after biological phosphorus removal or anaerobic selectors were implemented.

Full-Scale Plants – Effect on Dewatering



WRF, HRSD, CWS, Denver Metro Project Goals

Develop a fundamental, mechanistic understanding of bioflocs and their transformations in anaerobic digestion and how bioflocs interact with conditioning chemicals to affect the critical outcomes of:

1. polymer demand;
2. cake solids;
3. capture efficiency.

Project Goals

Apply this knowledge to:

1. improve dewatering at plants, and address issues such as the impacts of BioP and co-digestion;
2. to be able to include prediction of dewatering in appropriate models such as BioWin and SUMO

WRF PSC and Program Manager, Christine Radke

Project Technical Advisory Committee

Charles Bott, Sudhir Murthy, John Novak

PI

Dr. Matthew Higgins
Bucknell University

Outreach and Communications

Stacy Passaro
Passaro Engineering, LLC

Utility Partners

1. Dr. Chris Wilson, HRSD
2. J. McQuarrie/ L. Cavanagh, Denver
3. Dr. George Sprouse, MCES
4. Peter Schauer, CWS
5. Christine DeBarbadillo, DC Water
6. Kamlesh Patel, Chicago
7. Patrick Wescott, Green Bay MSD
8. Ester Rus Perez, Thames Water
9. Heri Bustamante, Sydney Water

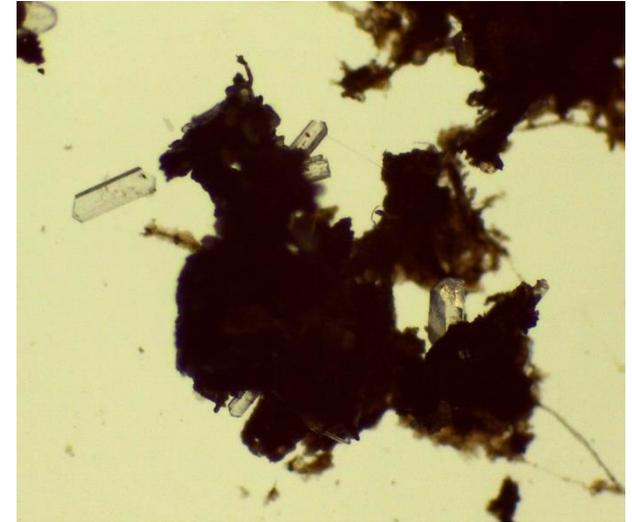
Consultant Partners

1. Dr. Wendell Kunjar, Hazen & Sawyer
2. Dr. JB Neethling, HDR
3. Dr. Sam Jeyanayagam, CH2M
4. Dr. Chris Muller, Brown & Caldwell
5. Dr. G. Rajagopalan, Kennedy/Jenks
6. Dr. Mo Abu-Orf, Hazen & Sawyer

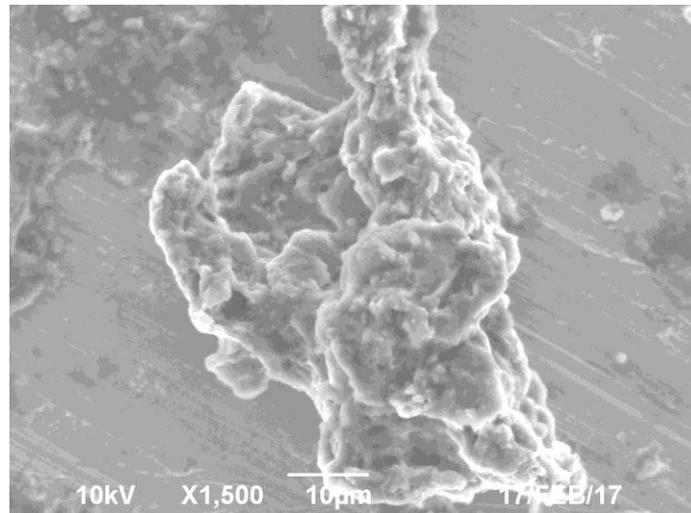
Solution Components

Components in Approximate Order by Mass

1. Water
2. Organics
 - A. Exocellular Polymeric Substances (EPS)
 - B. Cell Debris
 - C. Microbes
 - D. Miscellaneous other organics
3. Inorganics
 - A. Grit/Sand
 - B. Precipitates
 - C. Salts

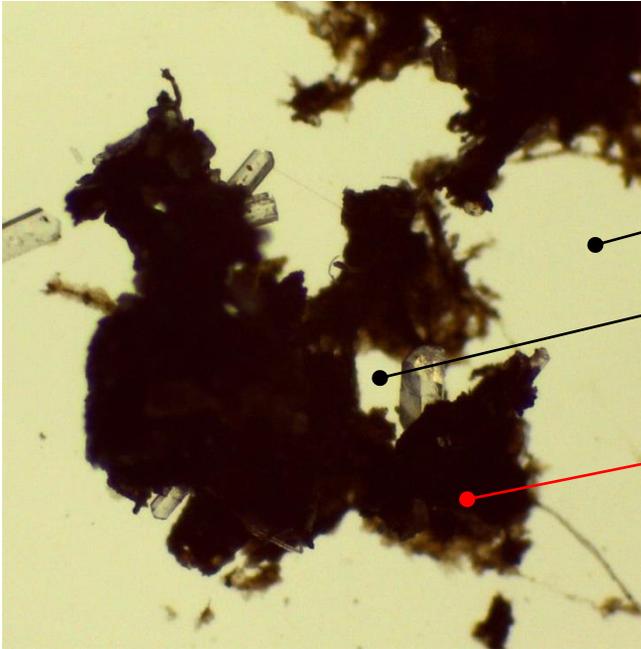


~100x



~1500x

Water: 95-99% of Solution Mass



~100x

Types of Water

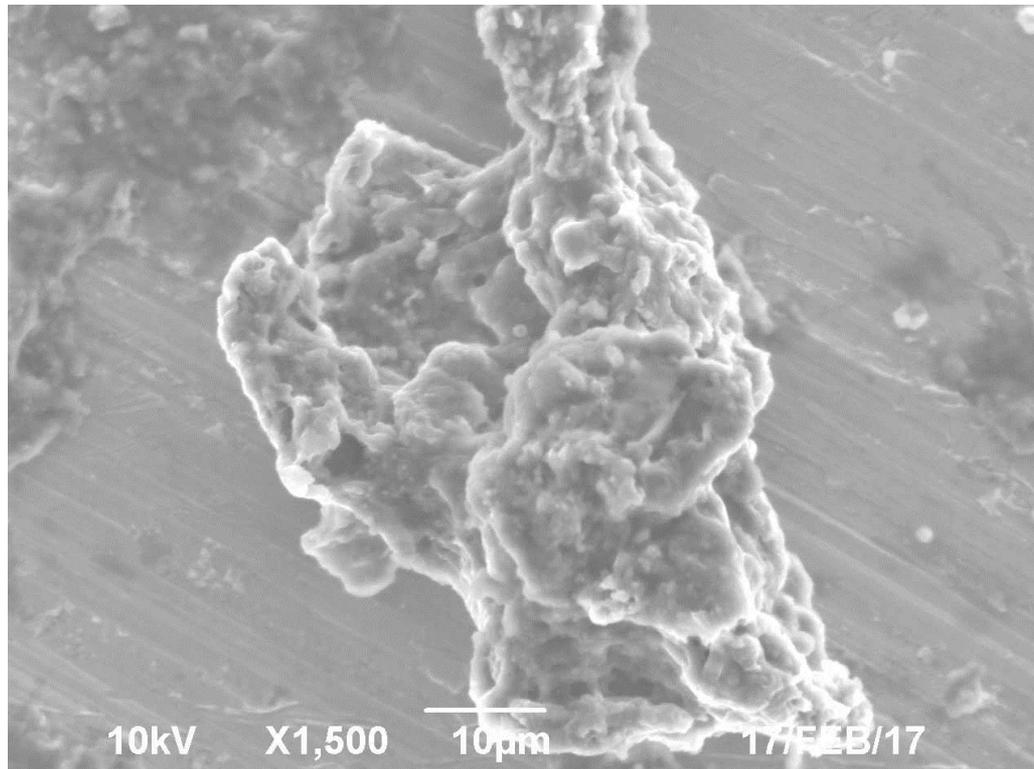
1. “Free” Water – water in bulk solution
2. “Interstitial” Water – water trapped in between flocs, and in pockets of water
3. “Floc” Water - water that is trapped with the floc

*“Only free water can be separated during mechanical dewatering.”
Dichtl and Kopp, 2000*

Biofloc Composition

Bioflocs

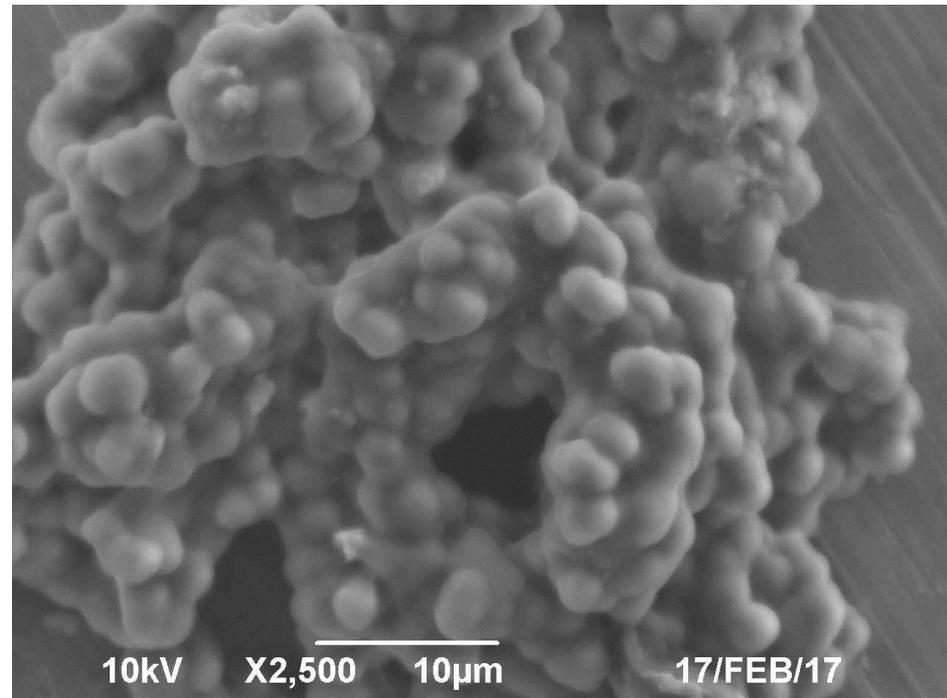
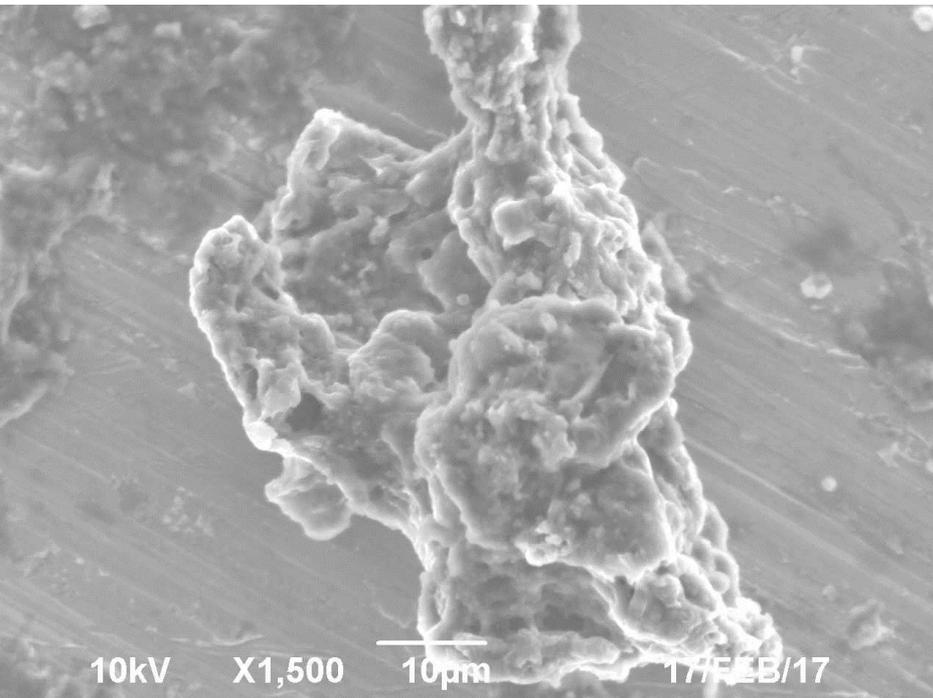
1. EPS – exocellular polymeric substances
2. Microbes
3. Grit, precipitates, misc. inorganics



EPS = Exocellular Polymeric Substances

EPS

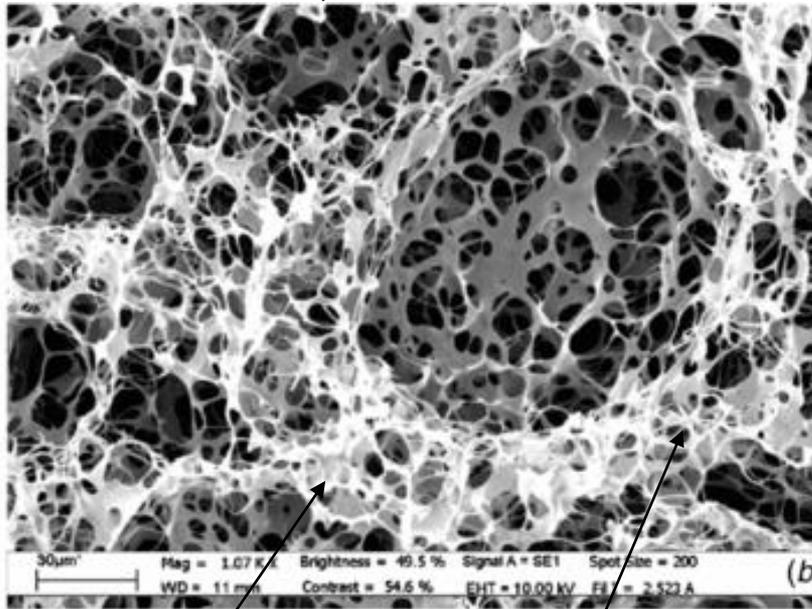
1. Proteins, Polysaccharides, Cell Debris
2. Have lots of negatively charged functional groups
3. Forms a gel within which microbes, water, and inorganics are embedded



EPS Characteristics – Trapped Water

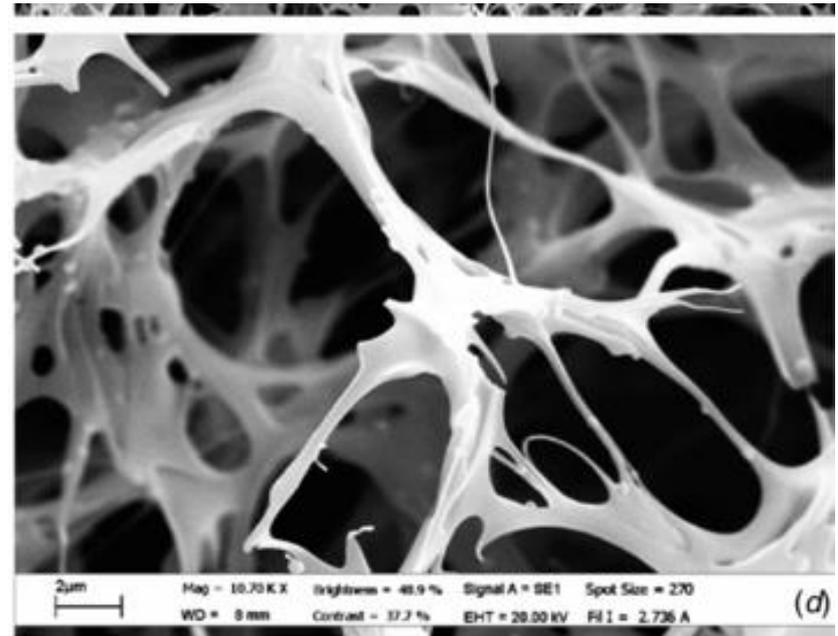
SEM of Gelatin Gel Network, (from De Colli et al., 2012)

1,000x



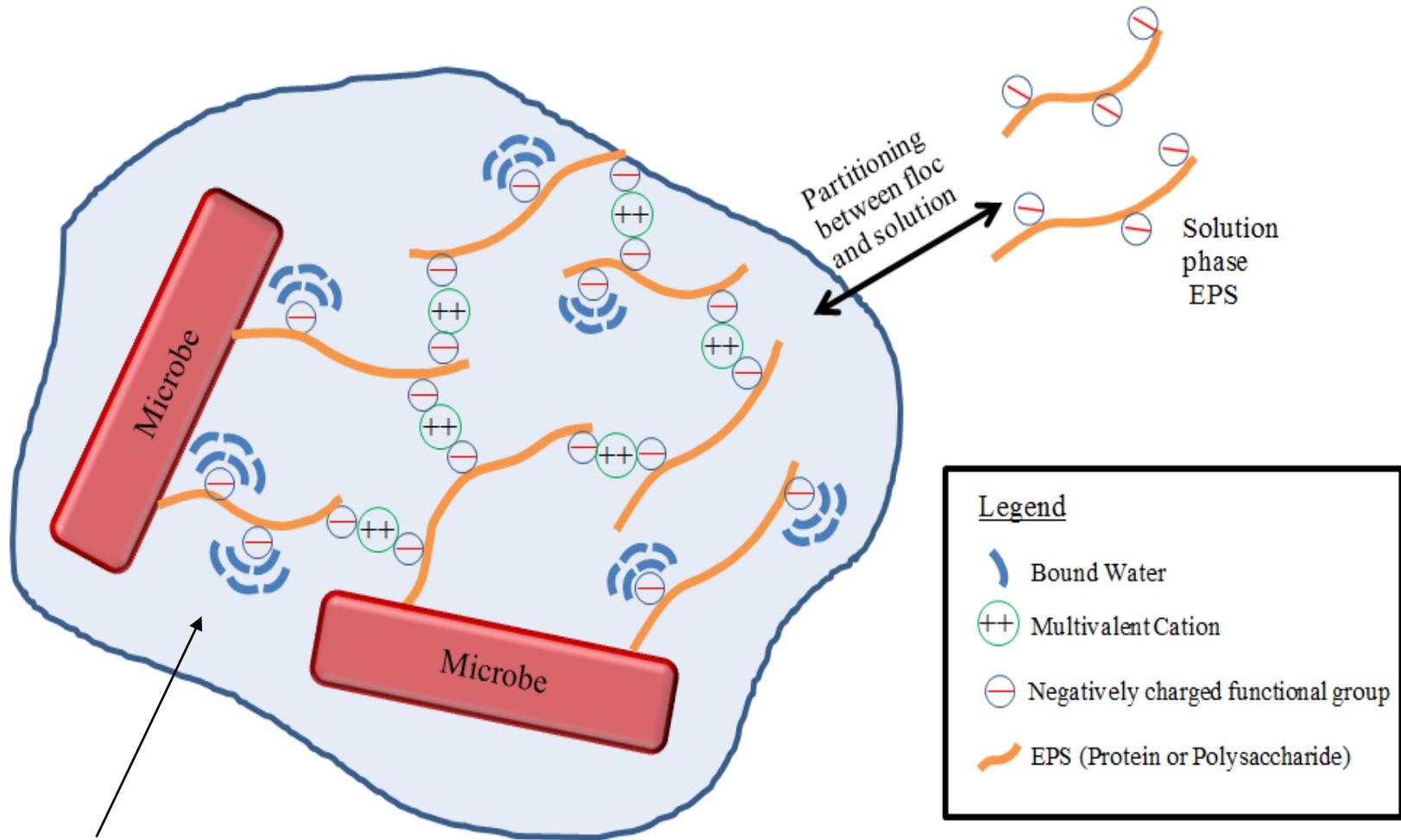
Gel “Fibers”
or EPS

10,000x



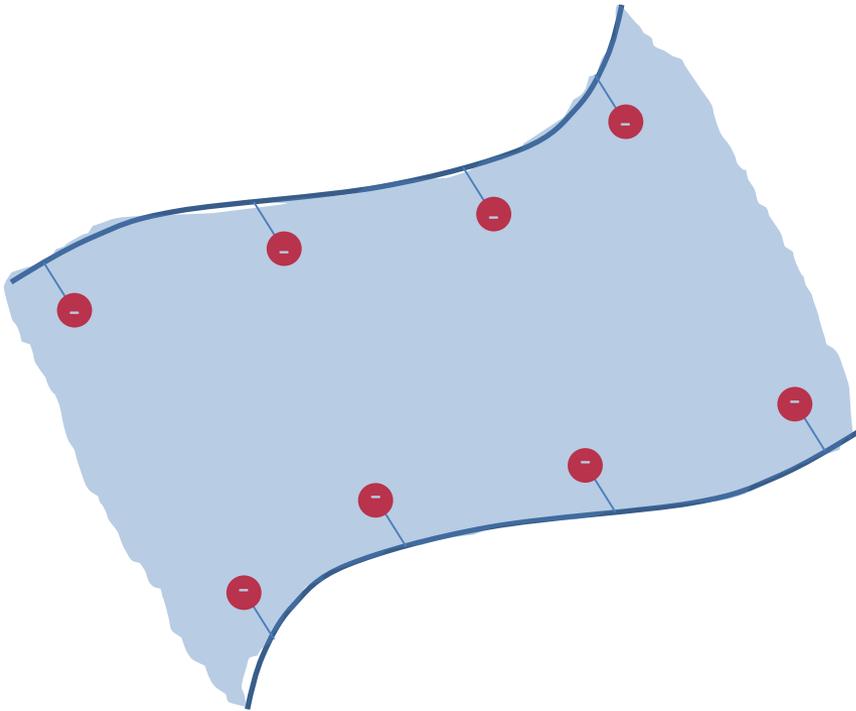
Water in cavities/pores is “trapped” water, has properties of free water, and can move within pores.
This is why Jello wiggles/deforms in response to stress

Working Hypothesis – Biofloc Model

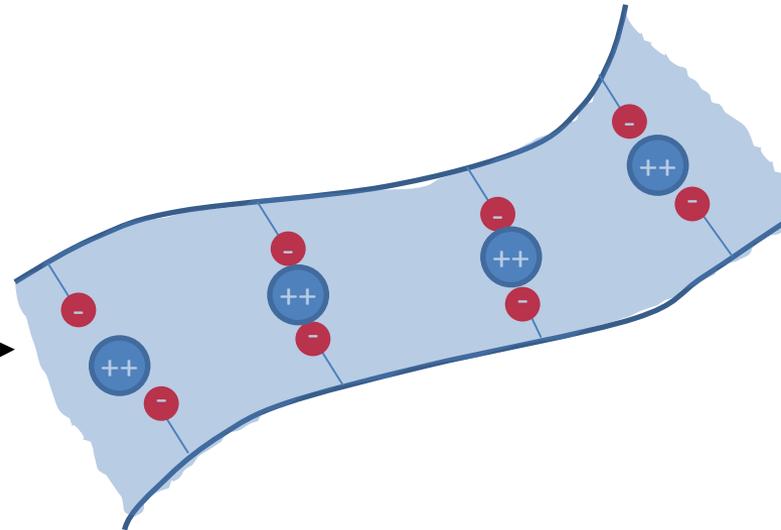


Divalent Cation Bridges Reduce “Pocket” sizes

Low Multivalent Cations

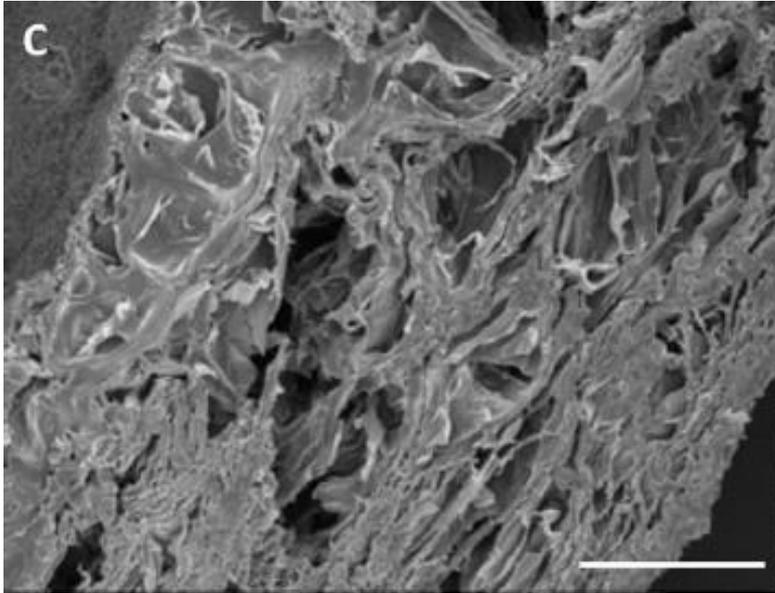


With Multivalent Cations

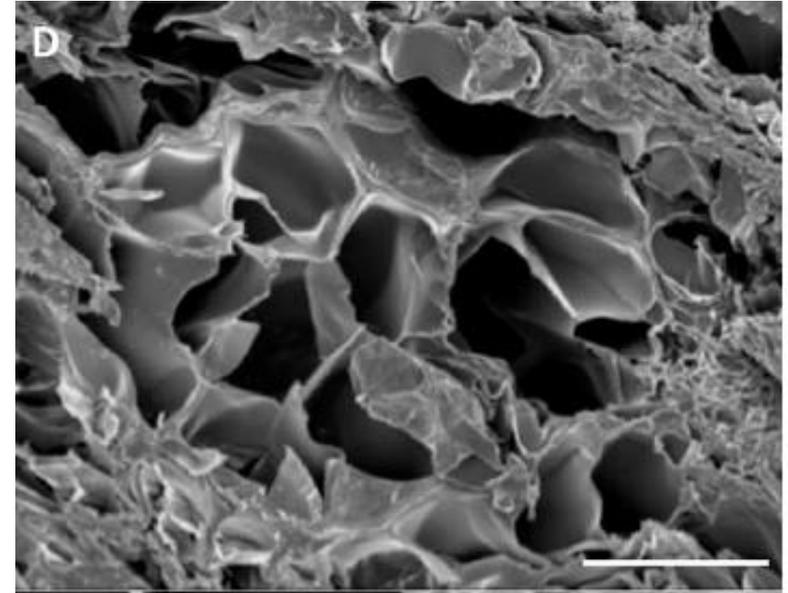


Divalent Cation Bridges Reduce “Pocket” sizes

Gel made with Ca^{2+}



Gel made without Ca^{2+}



OPEN

Increasing Mechanical Strength of Gelatin Hydrogels by Divalent Metal Ion Removal

SUBJECT AREAS:
SOFT MATERIALS
BIOMATERIALS

Received
23 October 2013

Accepted
28 March 2014

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Are you familiar with Divalent
Cation Bridging Theory?

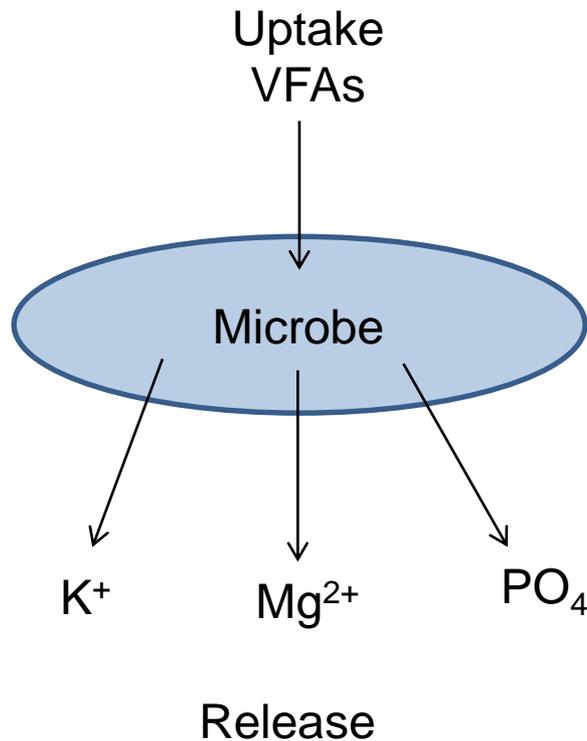
Hypothesis Summary

- Extent of dewatering is determined by water trapped in the floc
- EPS has negatively charged functional groups and gel-like properties which bind and trap water in the floc
- Divalent cations bridge negatively charged functional groups, displacing bound water and reducing trapped floc water

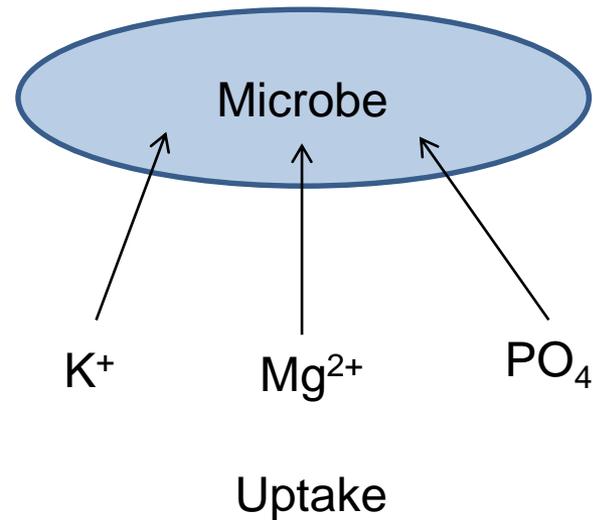
Biological Phosphorus Removal

Cycle Microbes and Feed through Anaerobic and Aerobic Periods

Anaerobic Phase



Aerobic Phase



Hypothesis for BioP Related to Biofloc Model

- Biological phosphorus removal results in a significant increase in digester PO_4^{3-} concentration.
- Divalent cations are sequestered by PO_4^{3-} species making them unavailable for biofloc formation and divalent cation bridging, resulting in a deterioration in floc properties and subsequent dewaterability.

Why Would BioP Impact Dewatering?

- Microbes release PO_4 , K^+ , Mg^{2+} in digester
- PO_4 can complex and precipitate Ca^{2+} and Mg^{2+} , making it unavailable for floc formation

Complexes	$\log(\beta\text{-values})^1$
$\text{Ca}^{2+} + \text{PO}_4^{3-} \leftrightarrow \text{CaPO}_4^-$	6.5
$\text{Ca}^{2+} + \text{HPO}_4^{2-} \leftrightarrow \text{CaHPO}_4^0$	15.1
$\text{Ca}^{2+} + \text{H}_2\text{PO}_4^{2-} \leftrightarrow \text{CaH}_2\text{PO}_4^+$	21.0
$\text{Mg}^{2+} + \text{PO}_4^{3-} \leftrightarrow \text{MgPO}_4^-$	4.8
$\text{Mg}^{2+} + \text{HPO}_4^{2-} \leftrightarrow \text{MgHPO}_4^0$	15.3
$\text{Mg}^{2+} + \text{H}_2\text{PO}_4^{2-} \leftrightarrow \text{MgH}_2\text{PO}_4^+$	20.0

Precipitates	K_{sp} values ¹
$\text{CaHPO}_{4(s)} \leftrightarrow \text{Ca}^{2+} + \text{HPO}_4^{2-}$	10^{-19}
$\text{Ca}_3(\text{PO}_4)_{2(s)} \leftrightarrow 3\text{Ca}^{2+} + 2\text{PO}_4^{2-}$	$10^{-28.7}$
$\text{Ca}_5(\text{OH})(\text{PO}_4)_{3(s)} \leftrightarrow 5\text{Ca}^{2+} + \text{OH}^- + 3\text{PO}_4^{2-}$	$10^{-58.2}$
$\text{MgHPO}_{4(s)} \leftrightarrow \text{Mg}^{2+} + \text{HPO}_4^{2-}$	$10^{-18.2}$
$\text{Mg}_3(\text{PO}_4)_{2(s)} \leftrightarrow 3\text{Mg}^{2+} + 2\text{PO}_4^{2-}$	$10^{-25.2}$
$\text{MgNH}_4\text{PO}_{4(s)} \leftrightarrow \text{Mg}^{2+} + \text{NH}_4 + \text{PO}_4^{2-}$	$10^{-18.2}$

Research Approach – Lab/Fundamentals

1. Sampling Survey to Evaluate Hypotheses:
 - a. Collected digestate from full scale anaerobic digesters with and without bioP
 - b. Operated lab digesters
 - c. Characterize the solids, solution chemistry and dewaterability with a defined laboratory dewatering protocol

Lab Protocol for Conditioning and Dewatering

Mix Polymer and Solids



Gravity Drainage



Centrifuge Cup

Pressure Applied Using Centrifuge

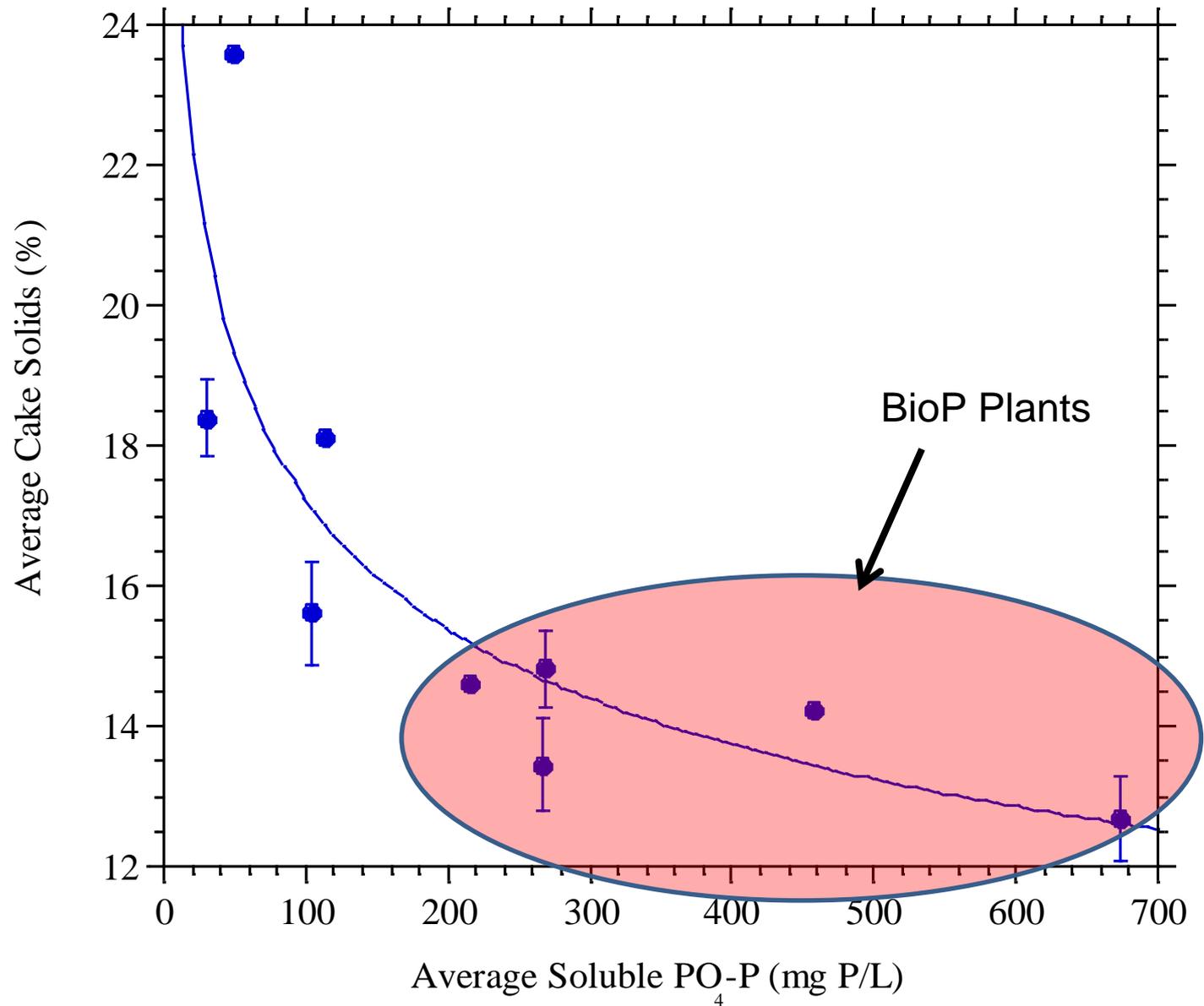
Filtrate Analyzed for TSS and other parameters



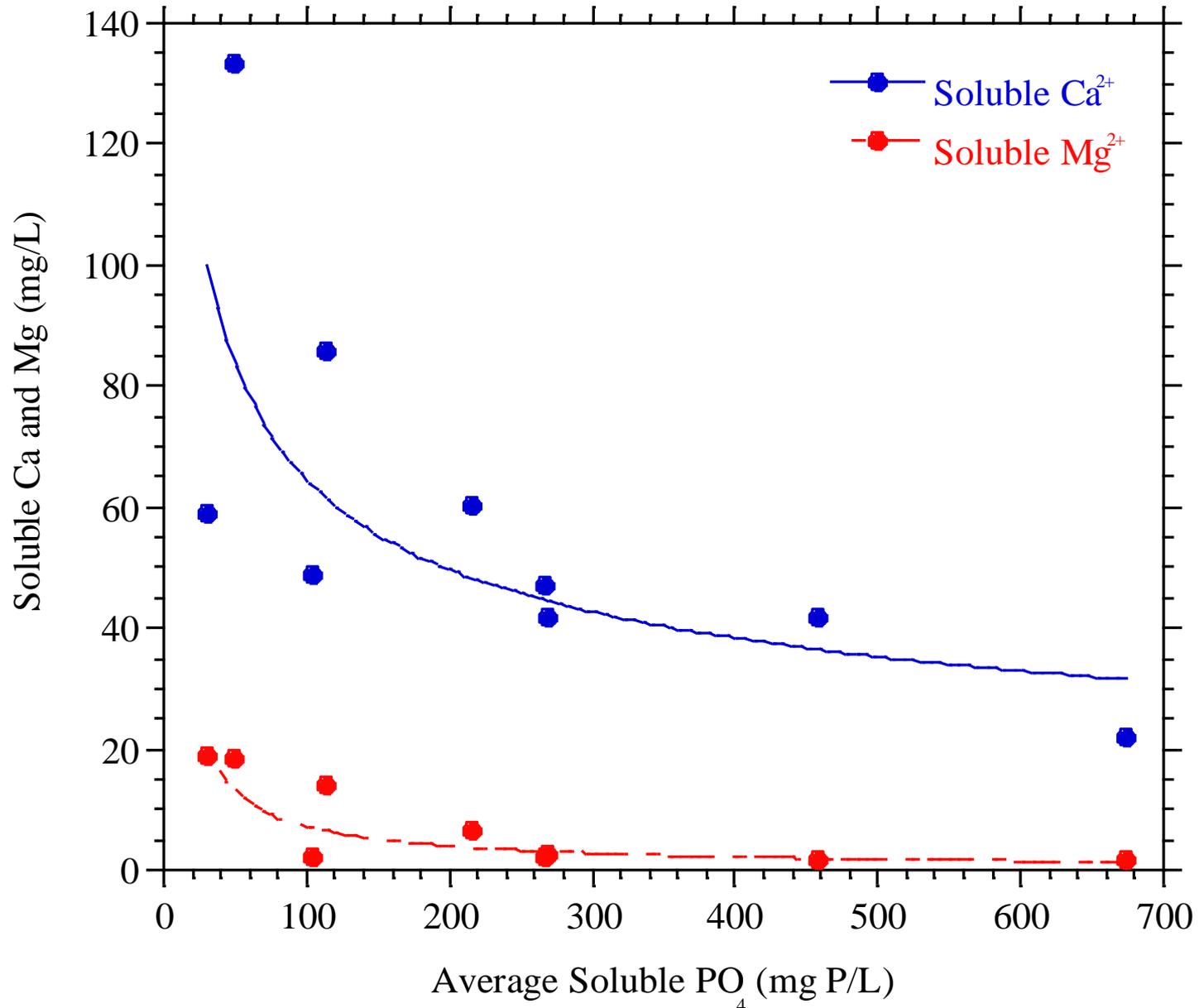
Measure CST



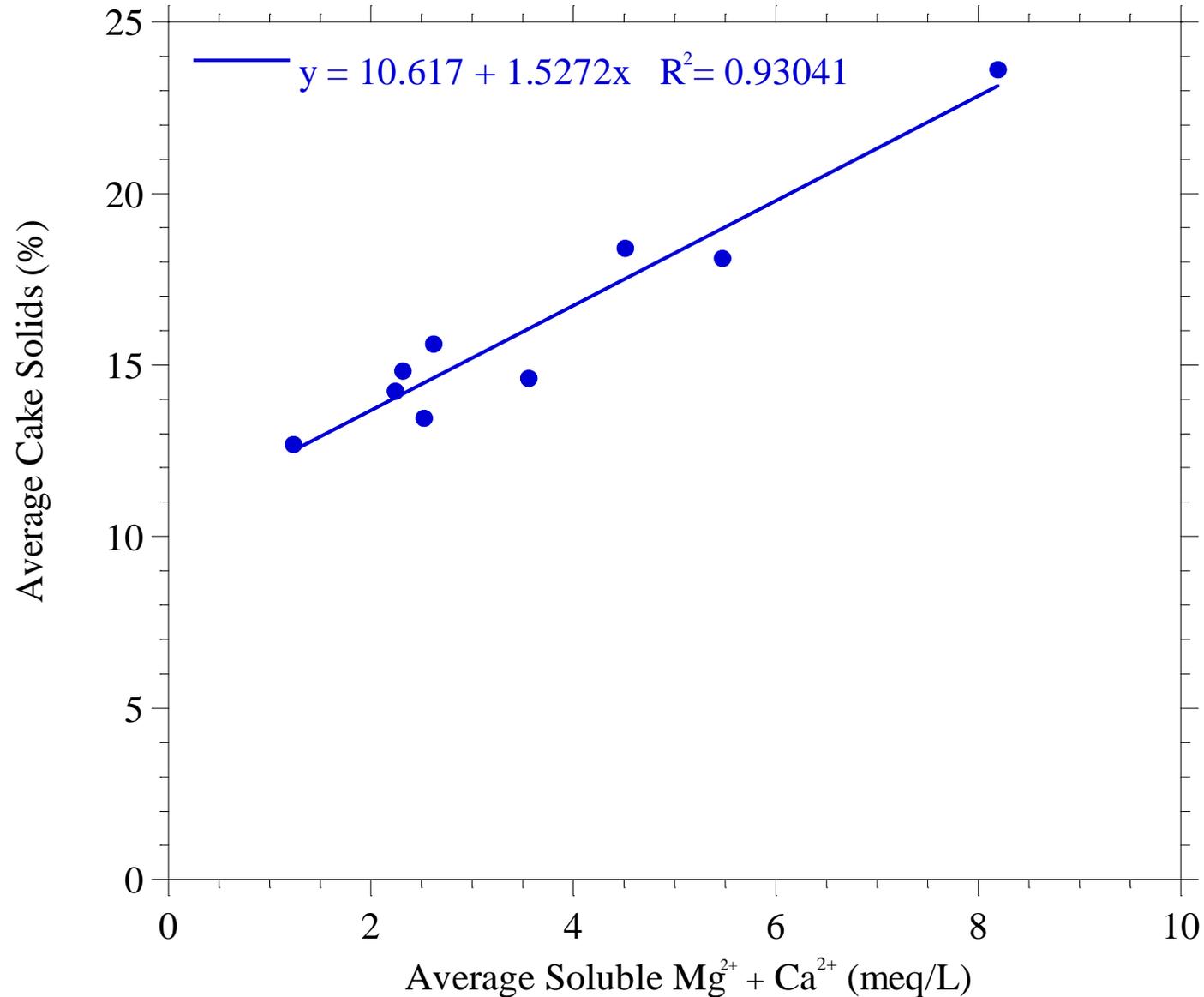
Results – Cake Solids vs PO_4



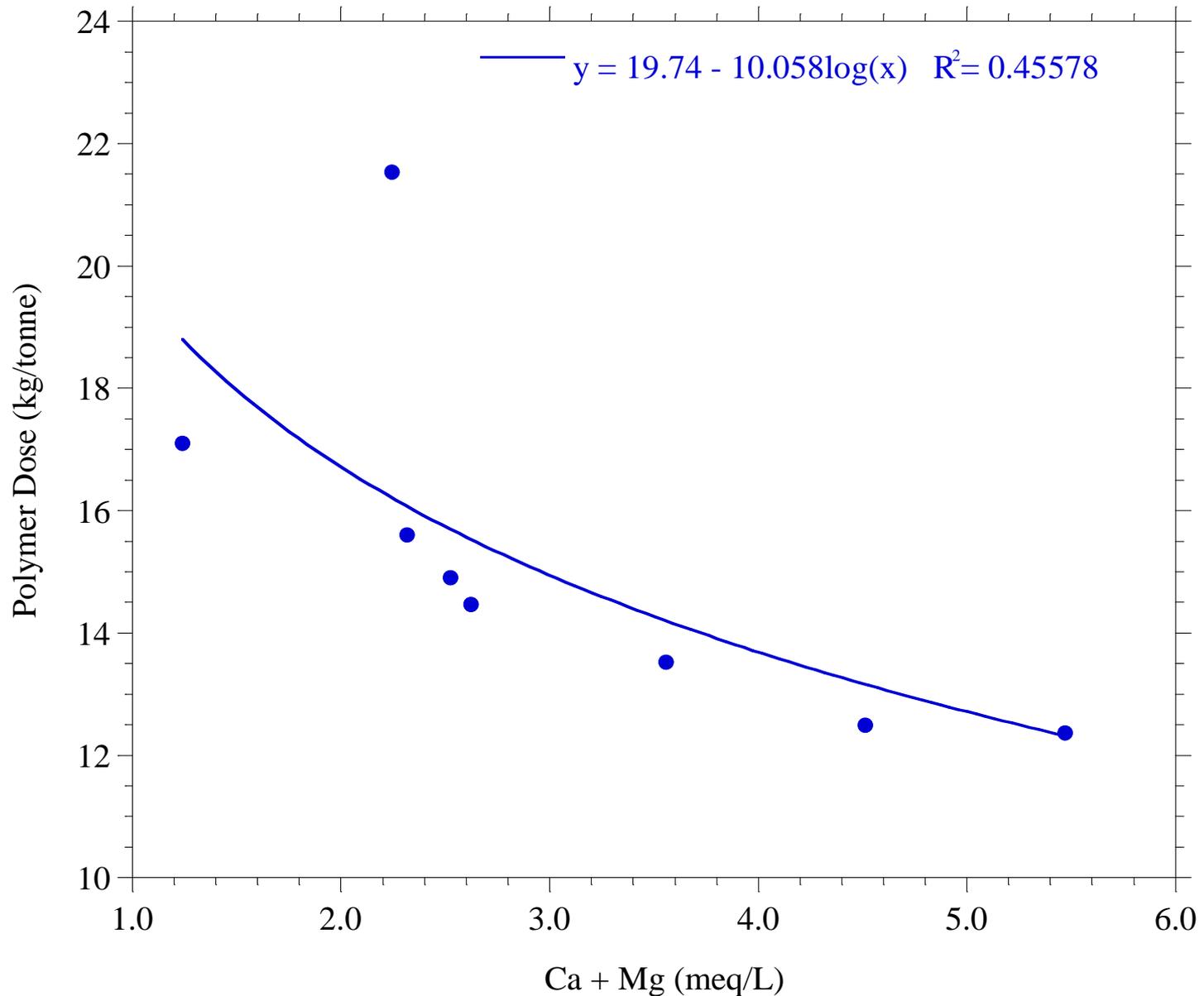
Soluble Ca^{2+} and Mg^{2+} vs PO_4



Results – Cake Solids vs Soluble ($\text{Ca}^{2+} + \text{Mg}^{2+}$)



Results – Polymer Demand vs Soluble ($\text{Ca}^{2+} + \text{Mg}^{2+}$)

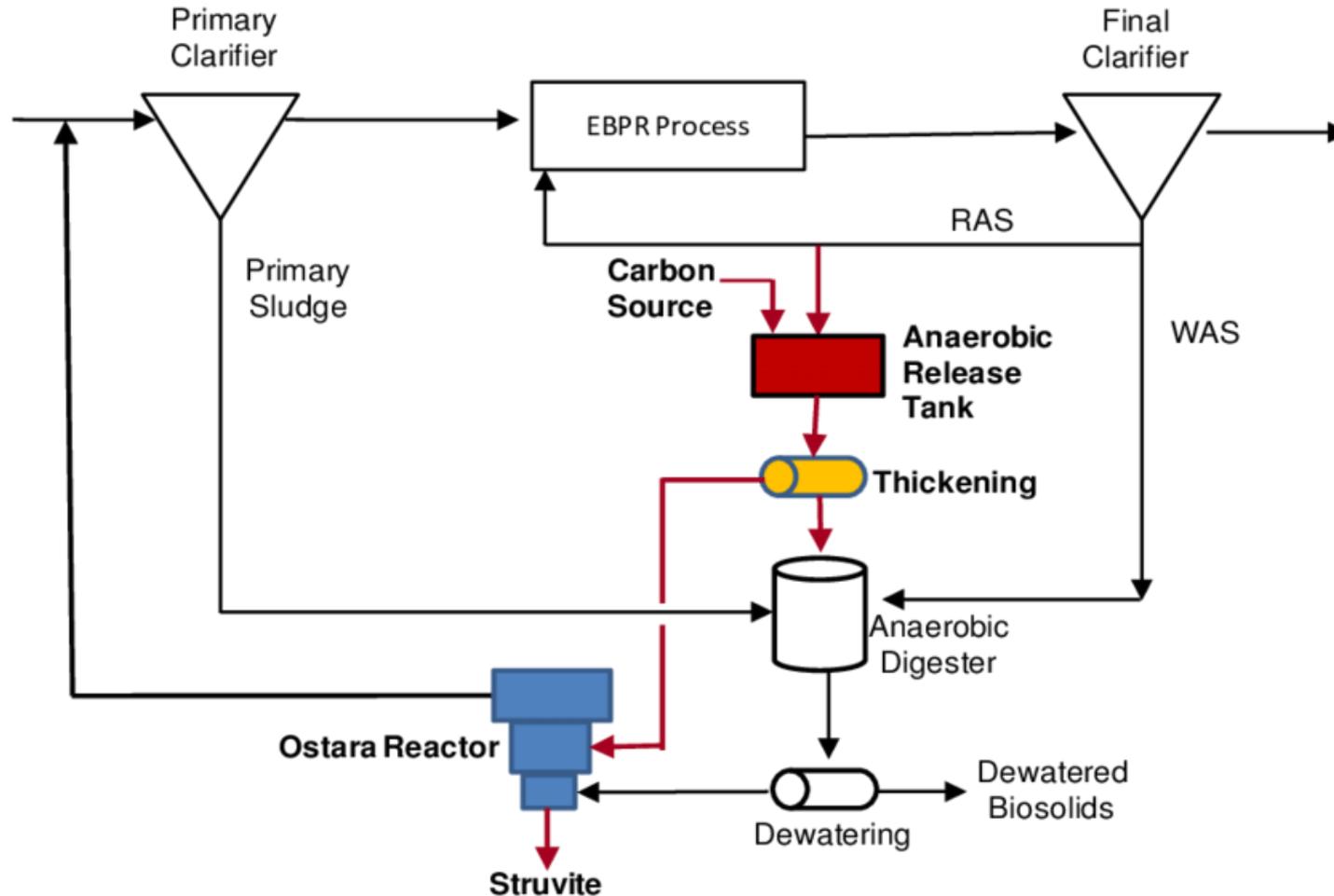


What can we do about the negative impacts?

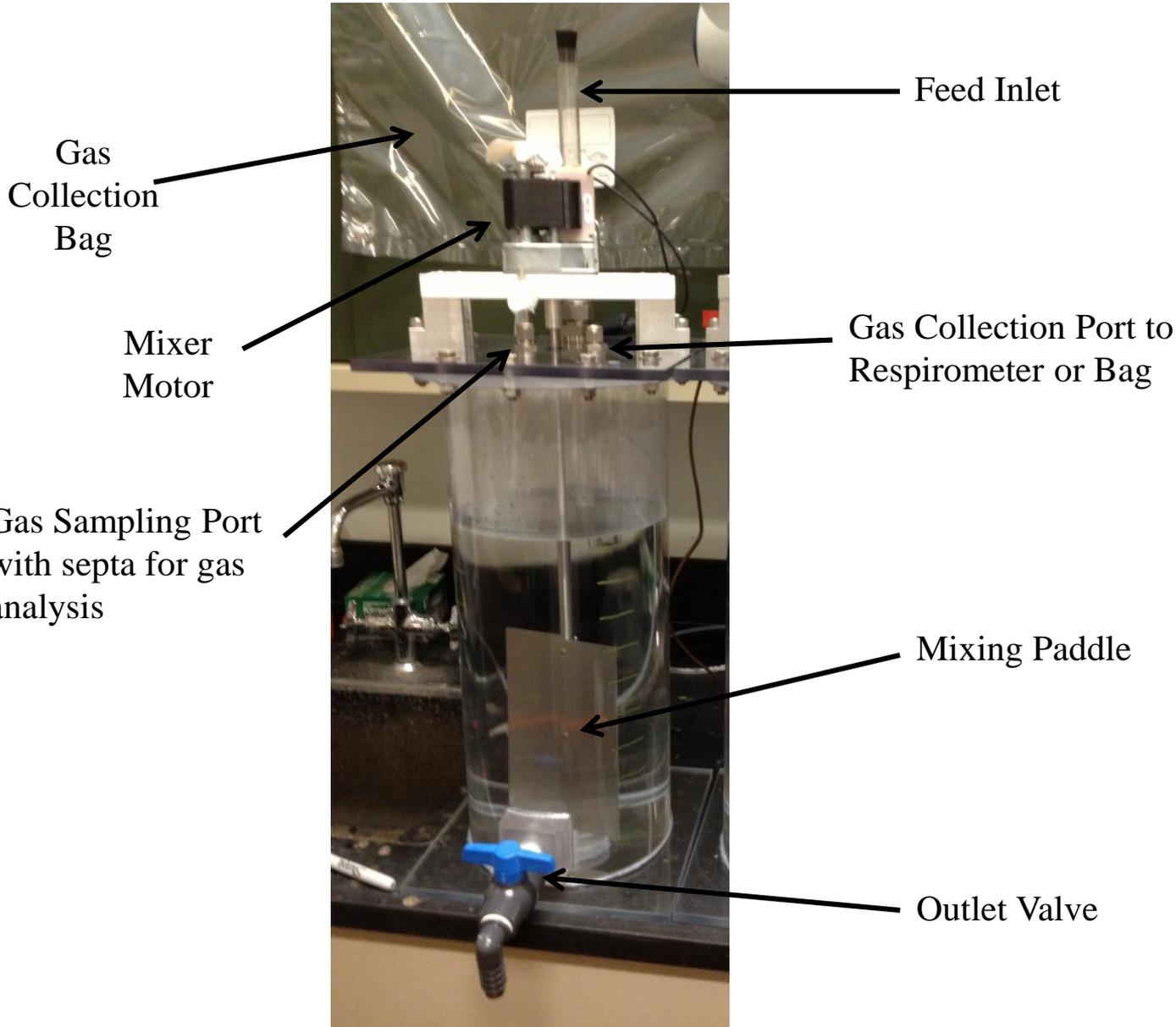
1. Don't do BioP, and use Fe or Al for P-removal
2. Phosphorus Stripping Prior to Digestion
3. Metal Addition (Fe, Ca, Mg, Al)
4. AirPrex

P-Stripping (WASSTRIP)

Ferment Biomass to release P, then thicken WAS before digestion

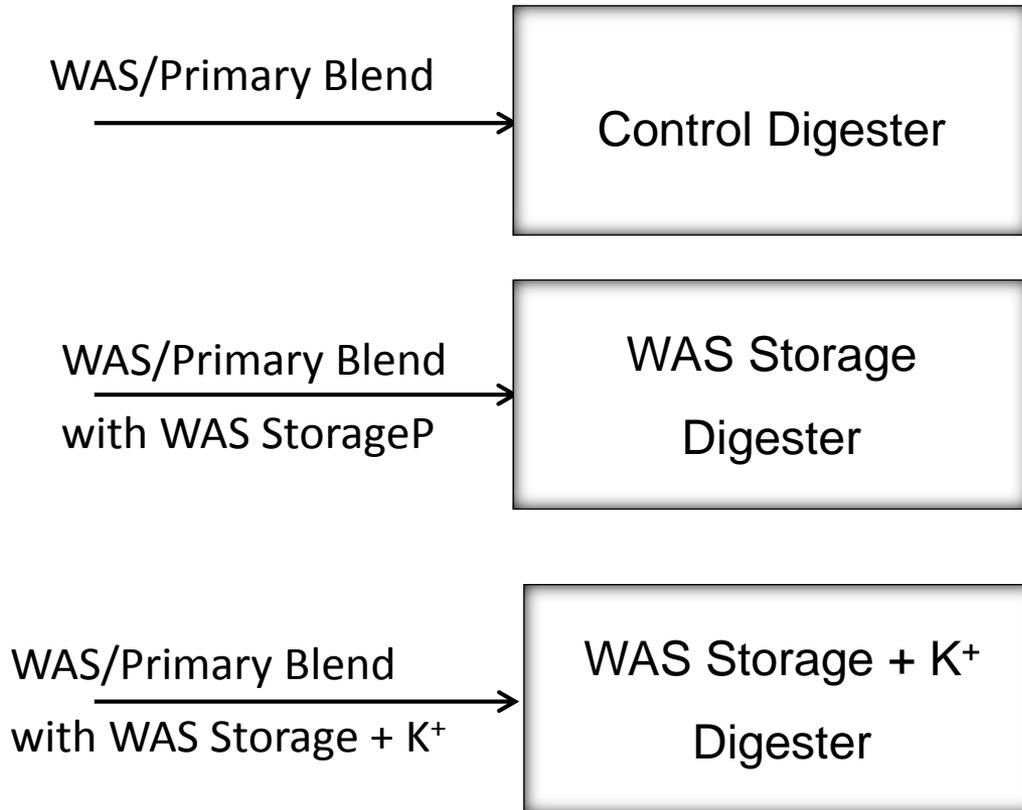


Lab Digester Reactors

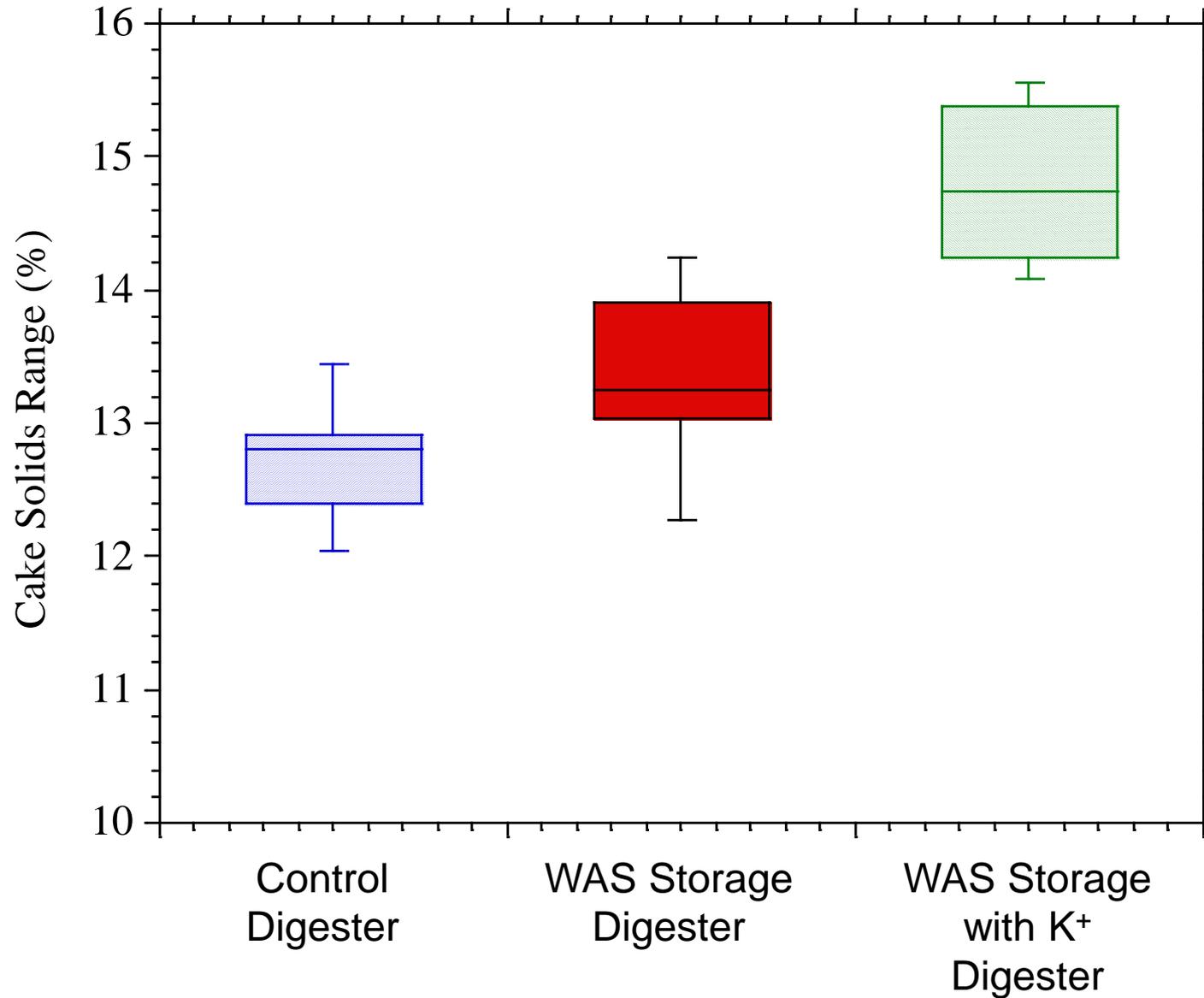


Volume = 10 L
SRT = 20 days
 $Q = 0.500 \text{ L/d}$

Reactor Feeds



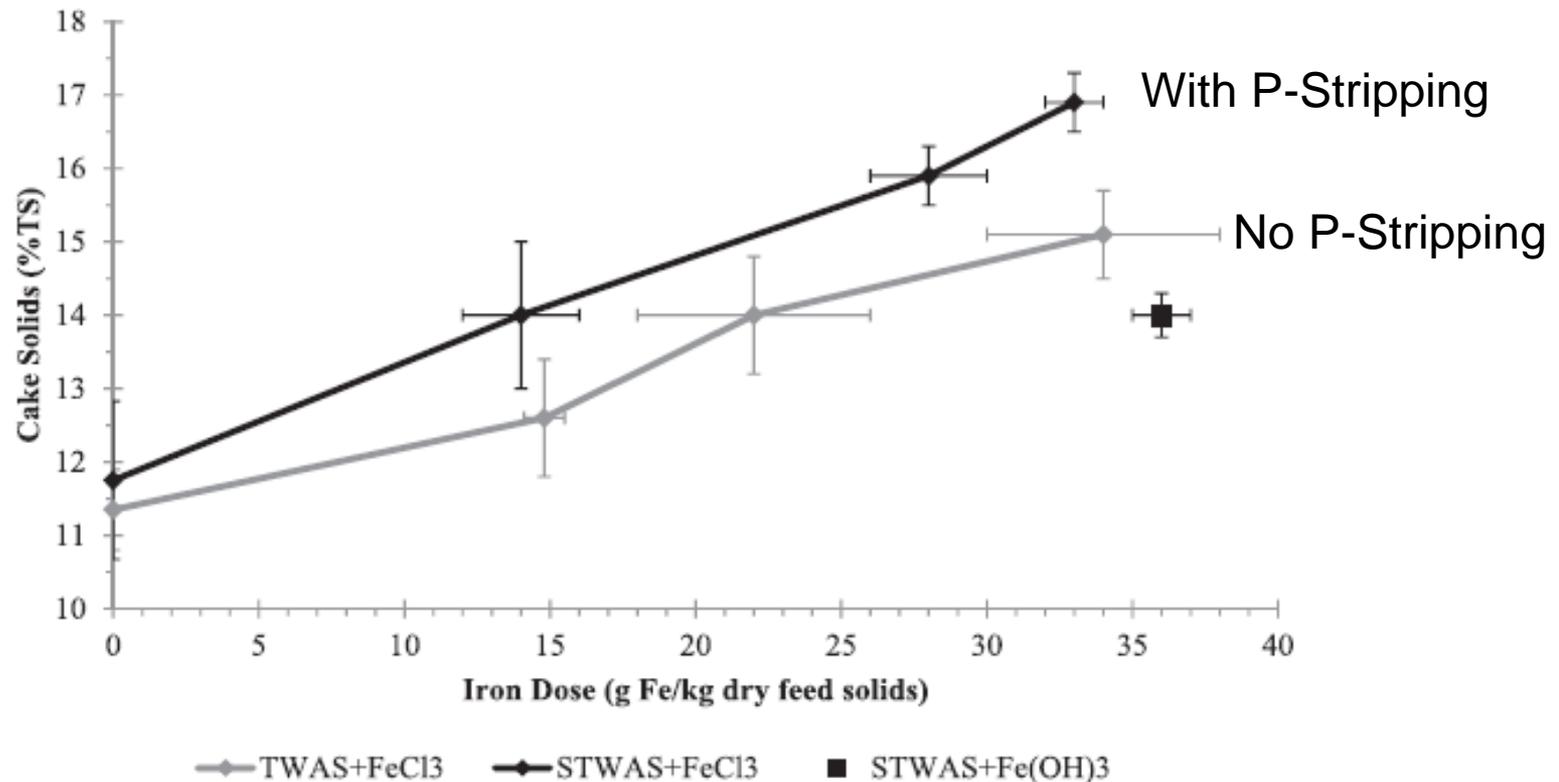
Results – Cake Solids for Digesters



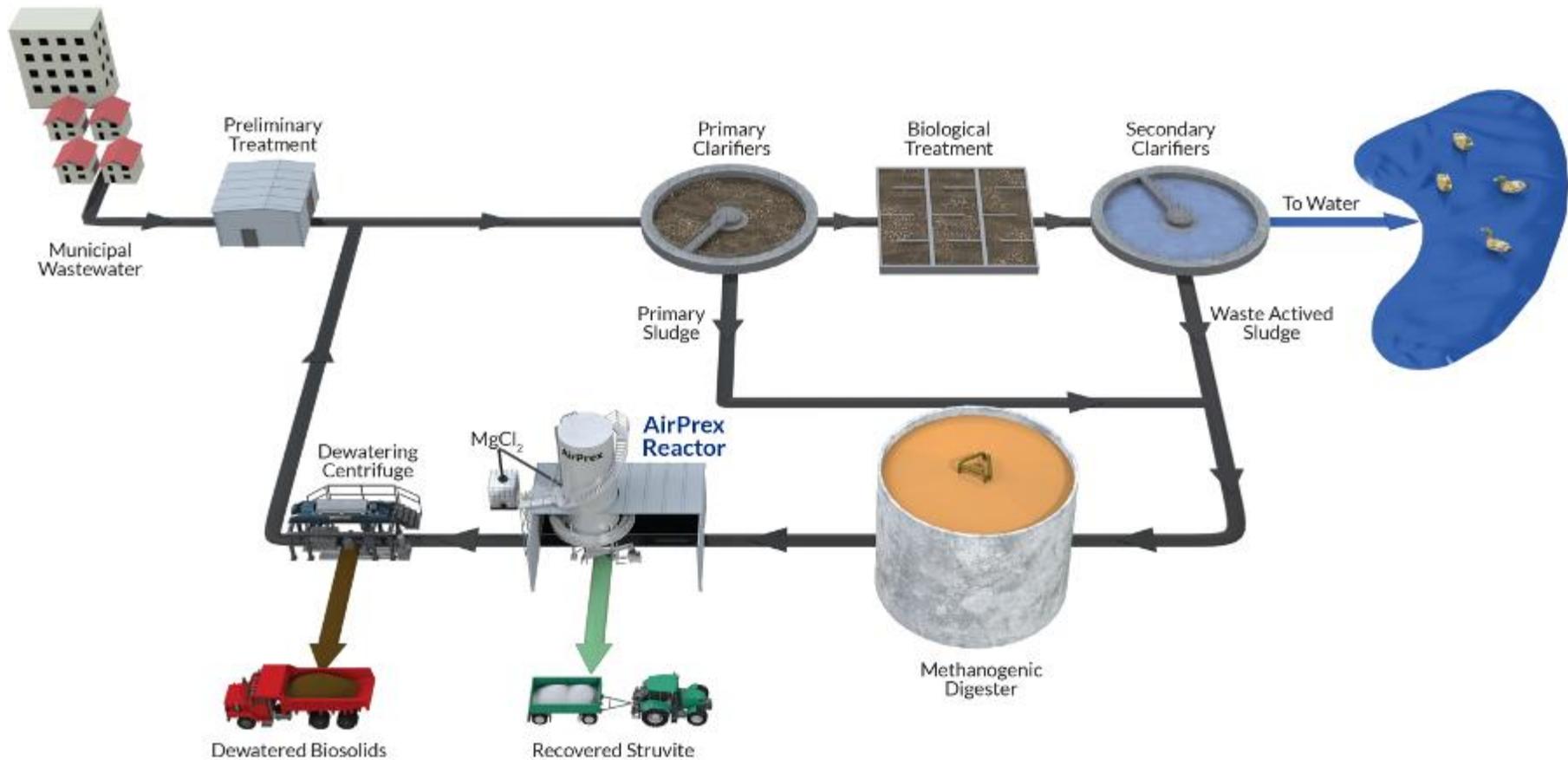
Metal Addition

Source: Water Environment Research, 2016
**Investigations into Improving Dewaterability at a
Bio-P/Anaerobic Digestion Plant**

Rebecca Alm¹, Adam W. Sealock¹, Yabing Nollet¹, George Sprouse^{1*}

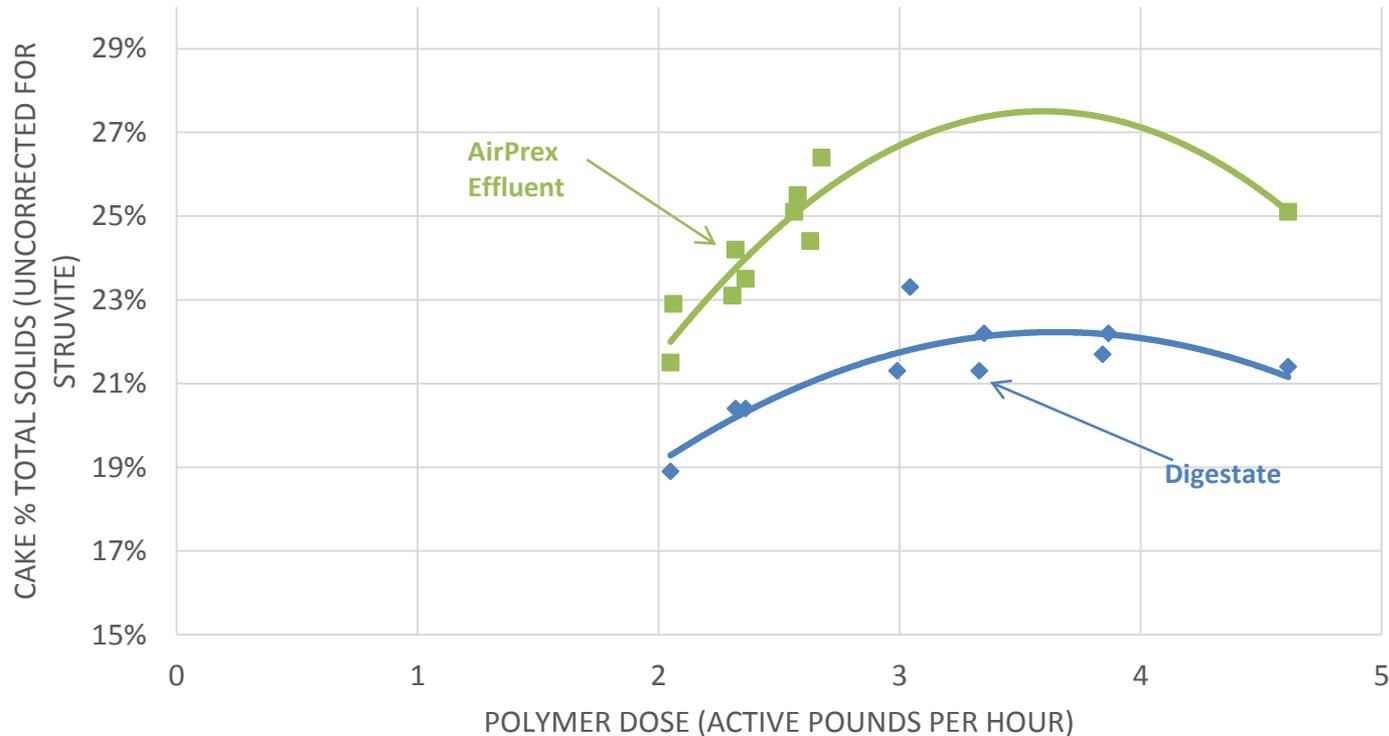


AirPrex



source: www.tpomag.com

AirPrex Pilot – Denver Metro



- 20 data points analyzed
- 8.72% reduction in wet tons hauled
- **17.61% decrease in polymer consumption**

Source: Wisdom et al., WEF Residuals and Biosolids Conference, 2017

Summary

- Biological phosphorus removal will increase phosphate in anaerobic digesters
- Phosphate impacts digester chemistry
- Chemistry is complex, phosphate binds multivalent cations (Ca, Mg, Fe, Al?) which increases water content of flocs
- Approaches to solving BioP issue revolve around reducing phosphate in digester, or increasing cation concentrations

Acknowledgements - Funding



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Denver Metro

Acknowledgements

Steven Beightol

Research Associate, Bucknell University

- performed the dewatering tests
- performed analytical testing and characterization
- compiled data

